INTRAMOLECULAR AZO COUPLING IN A NUMBER

OF 1-SUBSTITUTED 2-AMINOBENZIMIDAZOLES

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Benzimidazole-2-diazonium salts formed by the action of nitrosylsulfuric acid on 2-aminobenzimidazoles have extremely high activity and can enter into azo coupling at the 5 and 6 positions of the starting amine [1]. We have found that in the case of 1-aryl-2-aminobenzimidazoles (I), one observes, in addition to intermolecular azo coupling, intramolecular azo coupling, which proceeds at the ortho position of the Naryl radical and leads to the formation of dibenz [a,g]imidazo [2,1-c][1,2,4] triazines (II-V). The cyclization proceeds smoothly in concentrated H₂PO₄; the presence of concentrated H₂SO₄ promotes the formation of 5- and 6-azobenzimidazoles. A methyl group in the meta position of the N-aryl radical appreciably facilitates cyclization; in this case, two isomers (Va,b), which could be separated on plates with a thin layer of aluminum oxide in benzene -chloroform (1:2), are formed. The reaction is carried out by mixing (in the cold) 2 mmole of I [2], 5-7 ml of concentrated H₃PO₄, and 0.5 g (4 mmole) of solid nitrosylsulfuric acid [3]. The mixture is held at 20-25°C for 6 h and worked up as described in [1]. The triazines are isolated with a column filled with aluminum oxide; the bright-yellow fraction is washed out with chloroform. The structures of II-V were confirmed by the results of elementary analysis, IR spectroscopy, determination of the molecular weights by the Rast method, and by the observed analogy in the properties of II-V and benz[a]imidazo[2,1-c][1,2,4]triazines obtained from N-(o-aminophenyl)imidazoles [4]. Consequently, the concept of the structure of the previously obtained II as being a diazoamino compound [1] must be considered to be erroneous.

This method was used to obtain dibenz[a,g]imidazo[2,1-c][1,2,4]triazine (II, in 64% yield, mp 312-313°; $C_{13}H_8N_4$) and its 3-methyl (III, in 67% yield, mp 289-290°; $C_{14}H_{10}N_4$), 3-methoxy (IV, in 70% yield, mp 255-256°; $C_{14}H_{10}N_4O$), 2 (or 4)-methyl (Va, mp 308-309°; $C_{14}H_{10}N_4$), and 4 (or 2)-methyl (Vb, mp 288-289°; overall yield of Va and Vb 91%; $C_{14}H_{10}N_4$) derivatives.

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